

How To Apply the Complementarity and Coupling Theorems in MCR Methods: Practical Implementation and Application to the Rhodium-Catalyzed Hydroformylation

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ABSTRACT: Multivariate curve resolution techniques can be used in order to extract from spectroscopic data of chemical mixtures the contributions from the pure components, namely, their concentration profiles and their spectra. The curve resolution problem is by nature a matrix factorization problem, which suffers from the difficulty that the pure component factors are not unique. In chemometrics the so-called rotational ambiguity paraphrases the existence of numerous, feasible solutions. However, most of these solutions are not chemically meaningful. The rotational ambiguity can be reduced by adding additional information on the pure factors such as known pure component spectra or measured concentration profiles of the components. The complementarity and coupling theory (as developed in *J. Chemometrics* 2013 27, 106–116) provides a theoretical basis for exploiting such adscititious information in order to reduce the ambiguity. In this work the practical application of the complementarity and coupling theory is explained, a user-friendly MATLAB implementation is presented, and the techniques are applied to spectral data from the rhodium-catalyzed hydroformylation process.

KEYWORDS: spectral recovery, factor analysis, multivariate curve resolution, complementarity and coupling, rhodium-catalyzed hydroformylation process

1. INTRODUCTION

For consideration is a given chemical reaction system with several (potentially unknown) chemical components. Spectroscopic measurements on this system are assumed to result in a series of k spectra. Each spectrum is a vector with n absorbance values of the chemical mixture with respect to a fixed wavelength grid. This spectral data can be stored row-wise in a $k \times n$ matrix **D**.

The matrix formulation of the Lambert–Beer law says that ${\bf D}$ has a factorization

$$\mathbf{D}_{k\times n} = \mathbf{C}_{k\times s} \mathbf{A}_{k\times s} \mathbf{$$

where the concentration factor $\mathbf{C} \in \mathbb{R}^{k \times s}$ is a nonnegative matrix which contains column-wise the concentration profiles

of the *s* pure components with respect to the given time grid. The spectral factor $\mathbf{A} \in \mathbb{R}^{s \times n}$ contains row-wise the pure component spectra. Nonlinearities and measurement errors can be taken into account by adding a small error matrix $\mathbf{E} \in \mathbb{R}^{k \times n}$ to the right-hand side of (1).

In chemical applications only the spectral data matrix \mathbf{D} is given and the unknown number *s* of independent components as well as the pure component factors \mathbf{C} and \mathbf{A} are to be determined. A serious obstacle for this reconstruction problem is the so-called rotational ambiguity. This means that \mathbf{D} usually

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has numerous factorizations into nonnegative matrices C and A. The problem is to select from this continuum of solutions the "one" chemically correct solution. The first systematic analysis of such sets of solutions was done by Lawton and Sylvestre¹ in 1971 for a two-component system. Up to now, vast literature has been devoted to the rotational ambiguity and its low-dimensional representation; see for example refs 2-7 and the references therein. However, a systematic analysis of the rotational ambiguity is not necessary for the determination of practically useful factorizations. Instead approximation methods have been developed, which belong to the multivariate curve resolution (MCR) techniques or to the self-modeling curve resolution (SMCR) methods; see section 2. Some of these methods are available in software form such as the popular MCR-ALS toolbox for multivariate curve resolution problems.^{8,9} A further software which is specialized in the computation of the area of feasible solutions is the FAC-PACK toolbox.^{10,11}

1.1. Aim of This Work. Here, we are focusing on another approach to reduce the rotational ambiguity, namely, on the complementarity and coupling theory.¹² This theory allows one to formulate restrictions on the feasible concentration profiles if information on the spectra is available and vice versa. The complementarity and coupling theory has a solid mathematical foundation and can be formulated in terms of linear and affine linear subspaces to which certain concentration profiles and spectra are restricted. The mathematical argumentation is to some extent related to the duality theory by Rajkó.¹³

In this work we show how the complementarity and coupling theory can practically be applied to spectroscopic data. User-friendly MATLAB code is presented which can be applied to spectral data matrices **D** as introduced previously. Finally, our techniques and program codes are applied to a series of k = 2641 spectra, each with n = 664 wavenumbers, from the hydroformylation of 3,3-dimethyl-1-butene with a rhodium/tri(2,4-di-*tert*-butylphenyl)phosphite catalyst in *n*-hexane. For this example problem those parameters are determined which are associated with feasible nonnegative solutions.

2. SPECTRAL RECOVERY PROBLEM

For a given spectral data matrix $\mathbf{D} \in \mathbb{R}^{k \times n}$, the spectral recovery problem encompasses the computation of (1) the number of independent components *s* and (2) the nonnegative matrices **C** and **A** with $\mathbf{D} \approx \mathbf{CA}$.

The most established approach to compute *s* and to compute the factors **C** and **A** is the singular value decomposition (SVD) of **D**.¹⁴ The SVD reads $\mathbf{D} = \mathbf{U} \mathbf{\Sigma} \mathbf{V}^{\mathrm{T}}$ with orthogonal matrices of left singular vectors $\mathbf{U} \in \mathbb{R}^{k \times k}$ and right singular vectors $\mathbf{V} \in \mathbb{R}^{n \times n}$. Further, $\boldsymbol{\Sigma}$ is a $k \times n$ diagonal matrix with the singular values on its diagonal and zeros elsewhere. For noisefree data the number of nonzero singular values equals the number of independent components s. For noisy data the numerical rank s of D is the number of singular values larger than a threshold value (a proper multiple of the machine precision). The first s left and right singular vectors serve as a low-dimensional basis for the representation of the factors C and A; see, e.g., refs 1, 15, and 16. In the following we use the same notation for the factors U, Σ , and V of the truncated SVD in which U and V contain only these s singular vectors corresponding to the largest singular values and Σ is the $s \times s$ diagonal matrix with these singular values on its diagonal. The direct way to construct C and A with respect to these bases of singular vectors is to introduce a regular matrix $\mathbf{T} \in \mathbb{R}^{s \times s}$ and its inverse in the form

$$D \approx U \Sigma V^T = \underbrace{U \Sigma T^{-1}}_{=:C} \underbrace{T V^T}_{=:A},$$
(2)

see, e.g., refs 17-19 on this approach. The introduction of T and its inverse implies a substantial reduction of the degrees of freedom for the factorization problem to compute C and A. The decisive point is that T has only s^2 matrix elements, but C and A together have (k + n)s matrix elements. Having reduced the degrees of freedom in this way, the so-called rotational ambiguity is still a difficult obstacle. Usually, a computed solution (C,A) is not unique and a continuum of solutions exists if only the nonnegativity constraints are applied; see, e.g., refs 5 and 20. Any regular $s \times s$ matrix **R** can be used to construct the new factors $\tilde{C} = \mathbf{CR}^{-1}$ and $\tilde{A} = \mathbf{RA}$. Obviously, these factors solve the factorization problem since $\mathbf{D}=\tilde{C}\tilde{A}.$ The new factors are called feasible if \tilde{C} and $\tilde{A} \geq 0$. Typically, numerous feasible solutions exist in the form of one continuum or multiple continua.³ Various techniques have been developed in order to choose proper solutions. For example one can introduce soft and hard constraints,^{21,22} kinetic models,^{22,23} or proper additional information on the system in order to compute an appropriate T and thus the factors C and A. Further valuable tools are the window and evolving factor analysis,^{15,24} the usage of uniqueness theorems,²⁵ and so on. The book series of ref 17 is an elaborate reference on the wide range of developments.

In this work we are also interested in the construction of such T which result in nonnegative factorizations. However, our focus is somewhat different. We want to analyze the *mutual* relation *of restrictions on the factor* A (for instance by given spectra) on the *restrictions for the feasible concentration profiles* and vice versa.

3. COMPLEMENTARITY AND COUPLING THEORY

The complementarity and coupling theory is a rigorous mathematical analysis of the mutual relation between the factors C and A: see ref 12. In the following we explicitly treat the case of known spectra and the resulting restrictions on the concentration profiles. However, the analysis also includes the case of given concentration profiles and the resulting restrictions on the spectra since C and A are interchangeable in the following theorems.

Next the notions "complementarity" and "coupling" are used in the following sense: If for example the first s_0 pure component spectra $\mathbf{A}(j,:)$, $j = 1, ..., s_0$, are known, then (1) the concentration profiles $\mathbf{C}(:,i)$ for the other components $i = s_0 +$ 1, ..., *s* are called *complementary* and (2) the concentration profiles $\mathbf{C}(:,j)$ for the components $j = 1, ..., s_0$ with the same indexes are called *coupled*.

3.1. Colon Notation. The colon notation allows a succinct representation of the complementarity and coupling theorems and their mathematical background from linear algebra. This notation allows one to extract single or multiple columns or rows from a matrix. For a matrix **M** the notation $\mathbf{M}(i_i,:)$ defines the *i*th row of **M** and $\mathbf{M}(i_1:i_2,:)$ is the submatrix of the rows i_1 to i_2 of **M**. Everything works similarly in transposed form; e.g., $\mathbf{M}(:,j)$ is the *j*th column of **M**. MATLAB also uses this notation.

3.2. Complementarity Theorem. The complementarity theory says that if a number of s_0 spectra of an *s*-component system is known, the complementary concentration profiles are

restricted to an $(s - s_0)$ -dimensional linear subspace. The most restrictive case (aside from the trivial case $s = s_0$ that all spectra are available) is then $s_0 = s - 1$. The latter case is treated by the next theorem.

Theorem 3.1 (simplified complementarity theorem): If all but one pure component spectra are known, then the concentration profile of the remaining pure component is uniquely determined aside from scaling.

The fundamental idea behind the complementarity theory is to analyze the impact of a given spectrum on T. This implies an effect on T^{-1} which can finally be expressed as a restriction on the factor C. The full complementarity theorem reads as follows; the proof is contained in ref 12.

Theorem 3.2 (complementarity theorem): If s_0 pure component spectra are known, then the remaining concentration profiles are elements of the $(s - s_0)$ -dimensional subspace,

$$C := \{ \mathbf{U} \Sigma y : y \in \mathbb{R}^{s}, \mathbf{T}(1 : s_{0}, :) y = 0 \}$$

with $T(1:s_{0}) = A(1:s_{0})V$.

In section 4 we explain how these mathematical statements can be transformed into a practically applicable form. To this end, MATLAB code is presented which can directly be applied to the spectroscopic data. However, the mathematical theory *strictly* holds for noise-free data and in the absence of any numerical rounding errors—but the results still hold *approximately* for experimental and slightly noisy data. For a more detailed discussion of the impact of noise, see section 5 and section 6 for an application to experimental data.

3.3. Coupling Theorem. As introduced in section 3 the coupling theory provides a relation between the *i*th pure component spectrum $A(i_{j}:)$ and the *i*th concentration profile C(:,i).

Theorem 3.3 (coupling theorem): If s_0 pure component spectra are known (without loss of generality, we assume these components to be indexed by $i = 1, ..., s_0$); then the coupled concentration profiles C(:,i) fulfill

$$C(:, i) \in C^{(i)}$$
 for $i = 1, ..., s_0$

Therein the $C^{(i)}$ are the $(s - s_0)$ -dimensional affine linear subspaces

$$C^{(i)} := \{ \mathbf{U} \Sigma y : y \in \mathbb{R}^{s}, \, \mathbf{T}(1 : s_{0}, :)y = e_{i} \}$$
(3)

with $T(1:s_0) = A(1:s_0) V.$

Each of the spaces $C^{(i)}$ is an affine linear space. It results (by left multiplication with $U\Sigma$) of the solution y of the underdetermined system of linear equations

$$T(1:s-1,:)y = e_i$$
 (4)

Therein $e_i \in \mathbb{R}^{s-1}$ is the *i*th standard basis vector, which is just the *i*th column of the $s \times s$ identity matrix. Since $\mathbf{T}(1:s_0,:)$ has the rank s_0 , its null space has the dimension $s - s_0$ and the space of solutions of (4) has the dimension $s - s_0$. See section 4.3 for the graphical visualization of the set of feasible profiles $\mathbf{C}(:,i)$.

3.4. Nonnegative Solutions. The restrictions of the complementarity and coupling theory are still to be combined with the nonnegativity constraint. While theorem 3.1 provides a unique solution (aside from scaling with a positive scaling parameter), the other theorems result in linear and affine linear subspaces including one or more degrees of freedom. Subsets of these subspaces are to be identified which contain only the

nonnegative concentration profiles. In the following we consider two types of restrictions:

(I) Explicit nonnegativeness: $C(:,i) \ge 0$ is additionally required for any concentration profile as predicted by the complementarity and coupling theory.

(II) <u>Consistency</u>: The rank-reduced spectral data matrix $\mathbf{D} - \mathbf{C}(:,i) \overline{\mathbf{A}(i,:)}$, which represents the spectral data matrix \mathbf{D} after subtraction of the *i*th pure component, must again be nonnegative.

For more details see ref 12.

3.5. Usefulness of the Complementarity and Coupling Theory. Multivariate curve resolution methods suffer from the rotational ambiguity. The extraction of the "true" solution is a difficult problem which can approximately be solved by introducing hard and soft models (regularizations). Often some additional knowledge on the factors is available.

The complementarity and coupling theory is a mathematically rigorous technique to exploit this knowledge for the computation of a proper factorization $\mathbf{D} = \mathbf{CA}$. Formally the complementarity and coupling theory can be understood as a hard model for the reduction of the rotational ambiguity. However, noisy data can result in problems if the truncated SVD $\mathbf{U\Sigma}(:,1:s) \mathbf{V}(:,1:s)^{\mathrm{T}}$ is only a poor approximation of \mathbf{D} . Then $\|\mathbf{D} - \mathbf{U\Sigma}(:,1:s) \mathbf{V}(:,1:s)^{\mathrm{T}}\|_{F}$ is not small and the residual may contain unconsidered pure component information. See section 5 for more details.

4. PRACTICAL IMPLEMENTATION OF THE COMPLEMENTARITY AND COUPLING THEORY

In this section we give detailed guidance on how to apply the complementarity and coupling theory to spectral data matrices. The spectral data matrix is $\mathbf{D} \in \mathbb{R}^{k \times n}$, and we assume s_0 pure component spectra to be given. These spectra are inscribed row-wise into the matrix $\mathbf{A} \in \mathbb{R}^{s_0 \times n}$.

The program code is provided for the very popular MATLAB (MATrix LABoratory) numerical computing environment. Algorithms from numerical linear algebra are easily accessible in MATLAB as high-level language elements. With some additional effort the program code can be transferred to any other program language.

4.1. Initial Steps. The initial steps for the implementation of each algorithm is to compute an SVD of **D** (line 1 in each algorithm) and to ensure a proper orientation of the singular vectors (lines 2–7 in each algorithm). By testing $\max(\mathbf{V}(:,i)) \ge -\min(\mathbf{V}(:,i))$ and optional multiplication of the *i*th left and right singular vectors by -1, the singular vectors get an orientation which is numerically reproducible. Otherwise, some annoying sign ambiguity would interfuse the representation of the numerical results—especially if the complementarity and coupling theory is considered in the context of the computation of the area of feasible solutions (AFS); cf. ref 26. In line 8 the transformation **T** according to (2) is defined.

4.2. Implementation of the Complementarity Theorem. Algorithm 1 is an implementation of the simplified complementarity theorem 3.1. All but one spectrum are given; i.e., $s_0 = s - 1$. The null space of T is represented by the variable y (in line 9) and left-multiplication with U Σ results in the complementary concentration profile C(:,s) which is unique aside from scaling. Once again, the proper sign of C(:,s) is ensured by lines 11–13.

The implementation of the general complementarity theorem is more complicated. Algorithm 2 is an implementa-

Algorithm 1 Simplified complementarity.

Require: $D \in \mathbb{R}^{k \times n}$, $A \in \mathbb{R}^{(s-1) \times n}$, *s* **Ensure:** Complementary concentration c = C(:, s)1: [U,S,V] = svd(D); 2: for i=1:s if $-\min(V(:,i)) > \max(V(:,i))$ 3: 4: U(:,i) = -U(:,i);5: V(:,i) = -V(:,i);6: end: 7: end: 8: T = A*V(:,1:s); 9: y = null(T);10: $c = U^*S(:,1:s)^*y;$ 11: if -min(c) > max(c) 12: c = -c; 13: end; 14: plot(c);

Algorithm 2 Complementarity for $s_0 = s - 2$

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Require: $D \in \mathbb{R}^{k \times n}$, $A \in \mathbb{R}^{(s-2) \times n}$, s
Ensure: Complementary concentration profiles $C(:, [s-1, s])$ are plotted
1: [U,S,V] = svd(D);
2: for i=1:s
3: if $-\min(V(:,i)) > \max(V(:,i))$
4: $U(:,i) = -U(:,i);$
5: V(:,i) = -V(:,i);
6: end;
7: end;
8: $T = A^*V(:,1:s);$
9: Y = null(T);
10: [mi,i] = max(abs(Y(1,:)));
11: Y(:,[1 i]) = Y(:,[i 1]);
12: $Y(:,1) = Y(:,1)/Y(1,1);$
13: $Y(:,2) = Y(:,2)-Y(1,2)/Y(1,1)*Y(:,1);$
A suitable interval $[a, b]$ with maximal length $b - a$, so that all concentration profiles are non-
negative, is to be determined with the two restrictions (I./II.) from Section 3.4. If a minima
a and a maximal b have been determined, then an equidistant subdivision with $m = 20$ nodes
appears to be sufficient in order to plot a series of feasible solutions.
14: m = 20;
15° g = linspace(a, b, m):

16: plot(U*S(:,1:s)*(Y(:,1)*ones(1,m)+Y(:,2)*g));

tion of the case $s_0 = s - 2$; for the important case of an s = 3component system, this remaining option $s_0 = 1$ stands for a single given spectrum and is the only remaining nontrivial case. In the lines 10 and 11 the column vector of Y, whose first component has the largest modulus, is swapped to the first column of Y. The division by Y(1,1) guarantees that the resulting matrix Y fulfills Y(1,1) = 1. In line 13 the basis of the null space is modified in a way that Y(1,2) = 0. With these preparations and with a proper interval [a,b] which guarantees nonnegative concentration profiles, these profiles are plotted in line 16. There is only one such bounded interval [a,b], and a minimal a as well as a maximal b are to be computed so that the concentration profiles are nonnegative; cf. section 3.4. The two restrictions from section 3.4, namely, explicit nonnegativeness and consistency, are used to construct the two end points of the interval. Our construction of the first column of Y together with the Perron-Frobenius theory guarantee that this approach works properly.

In the lines 14-16 of algorithm 2 a plot of a series of m nonnegative concentration profiles is generated. Recommended values for m are 10, 15, or 20. All of this is demonstrated in section 6.2 for spectroscopic data from the rhodium-catalyzed hydroformylation.

4.3. Implementation of the Coupling Theorem. The initial steps in lines 1-8 of algorithm 3 are explained in section 4.1. The main difference compared with the implementation of the complementarity theorem is that the solution space is now an affine linear subspace.

Algorithm 3 is an implementation of the coupling theorem for $s_0 = s - 1$; i.e., all but one spectra of the pure components

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Algorithm 3 Coupling for s_0 = s - 1.
Require: D \in \mathbb{R}^{k \times n}, A \in \mathbb{R}^{(s-1) \times n}, s
Ensure: Coupled concentration profiles C(:, 1: s-1)
  1: [U.S.V] = svd(D);
  2: for i=1:s
       if -\min(V(:,i)) > \max(V(:,i))
 3.
           U(:,i) = -U(:,i);
          V(:,i) = -V(:,i);
        end;
 6.
 7: end:
 8: T = A*V(:,1:s);
 9: W = T\eye(s-1,s-1);
10: v = null(T)
     Each coupled concentration profile is an element of a one-dimensional affine subspace. For
    each i = 1, ..., s - 1 a proper interval [a_i, b_j] with maximal length b_j - a_j is to be determined
    which guarantees nonnegative concentration profiles.
11: i = 1;
                % i = 2; i = 3;
 12: m = 20;
13: g = linspace(a(i), b(i), m);
14: plot(U*S(:,1:s)*(W(:,i)*ones(1,m)+y*g));
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are known. In line 9 particular solutions for the $s - s_0$ inhomogeneous and underdetermined systems of linear equations

$$\mathbf{T}(1:s_0,:) \mathbf{W}(:,i) = e_i, \text{ for } i = 1, ..., s_0$$

are computed simultaneously. The *i*th column of W is a particular solution of the *i*th linear system. In line 10 the null space of T is computed. The null space is in general $(s - s_0)$ dimensional; for $s_0 = s - 1$ this linear space is one-dimensional. Hence, each solution has a single degree of freedom. For each *i*, $i = 1, ..., s_0$, a proper maximal interval [a(i),b(i)] is to be determined so that the two restrictions (I./II.) for the coupled concentration profile C(:,i) from section 3.4 are fulfilled. Restriction I is related with one end point of the interval, and restriction II is related with the other end point. The resulting profiles are plotted with respect to an equidistant subdivision of [a(i),b(i)] in lines 13 and 14 of algorithm 3.

The coupling theorem for general $s_0 \in \{1, ..., s - 1\}$ is implemented in a very similar way. Especially, lines 13 and 14 are to be changed as the higher dimensional null space of T requires a higher dimensional grid for the graphical representation of the feasible solutions.

5. NOISY AND EXPERIMENTAL DATA

The complementarity and coupling theorems 3.1-3.3 are formulated for noise-free data. A continuity argument shows that the results of the complementarity and coupling theorems still hold approximately for noisy or perturbed data if the signalto-noise ratio is large enough. However, if for a certain trace component the signal-to-noise ratio is very small, then the complementarity and coupling theory cannot be applied even if its pure component spectrum could be extracted by elaborated techniques. References on the extraction of pure component spectra for trace components with a very low signal-to-noise ratio and their successful confirmation, e.g., by density functional theory (DFT) computations, are ref 27 in section 4.4 or refs 28-30.

Next we discuss the influence of random and systematic noise on the results as well as its dependence on the ratio of total absorbance of a certain species to the level of noise. The effects also depend on the used spectroscopic technique.

5.1. Low-Rank Approximation by the SVD. A key step in spectral recovery techniques is the low-rank approximation of C and A by using only the s largest singular values and the associated s left and right singular vectors. If the reconstruction error D - U(:,1: s) $\Sigma(1: s, 1:s)$ V(:,1: s)^T is small, then multivariate curve resolution methods can work very well. However, in the presence of systematic noise and if the signalto-noise ratio for a specific component is not small, then the truncated SVD is not a reliable basis for the reconstruction of the correct solutions.^{19,29,31,32} In this case the low-rank representation T(:,1:s) = A(1:s,:)V cannot reconstruct the spectral data very well as the error $A(1:s,:) - A(1:s,:)VV^{T}$ is not small. Then an application of the complementarity and coupling theory cannot be recommended.

5.2. Trace Components with a Low Signal-to-Noise **Ratio.** If the noise level is relatively small and the signals of a trace component are of a size comparable to the noise level, i.e., the signal-to-noise level for this component is large, then a successful strategy is to work with z singular vectors in order to construct a number of s spectra with z > s. Then the matrix **T** in (2) is a rectangular $s \times z$ matrix and \mathbf{T}^{-1} is to be replaced by its pseudoinverse \mathbf{T}^+ ; see ref 29. For this more general situation the complementarity and coupling theory cannot be applied, since it has only been formulated for square matrices $\mathbf{T} \in \mathbb{R}^{s \times s}$.

5.3. Further Spectroscopic Techniques. Up to now we have successfully applied the complementarity and coupling theory to UV/vis and FT-IR data; see section 6 and refs 7, 12, and 26. Especially for UV/vis data the size and type of the noise is not interfering with the computational procedure. However, for FT-IR data a potential baseline correction is a critical step whose proper implementation is crucial for the subsequent computations. In principle the complementarity and coupling arguments appear to be useful building blocks for extracting pure component information if proper adscititious information on the chemical system is available. These techniques might be a part of a prospective automatic analysis of mixtures, cf. with the automatic analysis in X-ray powder diffraction.^{33,34}

6. APPLICATION TO THE RHODIUM-CATALYZED HYDROFORMYLATION PROCESS

In this section the numerical algorithms and program codes are applied to in situ FT-IR spectroscopic data from the rhodiumcatalyzed hydroformylation process. For the experimental details see ref 35. Within the spectral interval [1960,2120] cm^{-1} three dominant active species can be identified; two of the pure component spectra of the three components are known. These are ideal preconditions for the application of the complementarity and coupling theory.

6.1. Spectral Data and Two Pure Component Spectra. A series of k = 2641 spectra were taken from the hydroformylation of 3,3-dimethyl-1-butene with a rhodium/tri(2,4-di-*tert*-butylphenyl)phosphite catalyst ([Rh] = 3×10^{-4} mol/L) in *n*-hexane at 30 °C, p(CO) = 1.0 MPa, and $p(H_2) = 0.2$ MPa. Each spectrum is a vector with n = 664 absorbance values in the interval [1960,2120] cm⁻¹. Figure 1 shows 34 of these spectra. Within this spectral interval the reactant 3,3-dimethyl-1-butene as well as the hydrido and acyl rhodium complexes are the prevailing components; cf. ref 35. This statement is supported by the distribution of the singular values. The three largest singular values are characteristically larger than the remaining singular values which are close to zero. Thus, we set s = 3. Figure 2 shows the singular values together with the left and right singular vectors.

Two spectra of the reaction subsystem are known: The spectrum of the olefin 3,3-dimethyl-1-butene is available, and the spectrum of the hydrido complex is known. These two specta are shown in Figure 3.

6.2. Application of the Complementarity Theorem. The complementarity theorem can easily be applied. Two of



Figure 1. Selection of 34 of k = 2621 spectra for the $k = 2621 \times n = 664$ spectral data matrix **D** for the rhodium-catalyzed hydro-formylation process.

the three pure component spectra are available so that the simplified complementarity theorem 3.1 can be used. The concentration profile of the third component (acyl complex) is uniquely determined aside from scaling. algorithm 1 with s = 3 results in the concentration profile C(:,2) of the acyl complex; see Figure 4.

6.3. Application of the Coupling Theorem. Since all but one pure component spectra are available, algorithm 3 can directly be applied. Next we explain the computation of the concentration profile of the olefin. The profile for the hydrido complex can be computed similarly. After the initialization phase a particular solution W of the underdetermined system of inhomogeneous linear equations $TW(:,1) = (1,0)^T$ is computed; see line 9 in algorithm 3. In Figure 5 $U\Sigma W(:,1)$ is shown by the solid line. Then the null space of T is computed. Figure 5 shows $U\Sigma y$ as a broken line for a $y \neq 0$ from this null space. The affine linear space $C^{(1)}$ in theorem 3.3 is then spanned by all $C(:,1) = U\Sigma z$ with $z = W(:,1) + \gamma y$ and $\gamma \in \mathbb{R}$. Finally, a real interval for γ is to be determined so that C(:,1) satisfies the two restrictions (I and II) from section 3.4. For the given data we get $\gamma \in [a_1, b_1] = [1.19, 1.98]$. (For all other γ either C(:,1) has negative components or the rankreduced matrix $\mathbf{D} - \mathbf{C}(:,1)\mathbf{A}(i,:)$ has negative components.)

Figure 6 shows the resulting feasible concentration profiles for the olefin. Similarly, the feasible concentration profiles of the hydrido complex are also contained in a one-dimensional affine subspace. Together with the nonnegativity restrictions the remaining profiles are shown in Figure 7.

6.4. Complete Solution. We have shown previously that the complementarity and coupling theory with two given pure component spectra uniquely determines one concentration profile and restricts the concentration profiles of the remaining two components to one-dimensional affine subspaces. Thus, the complete factorization D = CA has still a single degree of freedom.

If some kinetic model is added (in the form of a soft constraint), then this remaining single degree of freedom can be removed; see refs 35 and 36 for the details. The resulting factors are shown in Figure 8.

7. CONCLUSION

The complementarity and coupling theory provides advantageous tools for multivariate curve resolution techniques in order to exploit the mutual dependence of the partial knowledge of one factor and the resulting restrictions on the



Figure 2. Singular value decomposition of data matrix D from rhodium-catalyzed hydroformylation: (left) the first three left singular vectors; (middle) the first 30 singular values in a logarithmic plot; (right) the first three right singular vectors.



Figure 3. Two known pure component spectra. The olefin (component 1) is shown by a blue line and the hydrido complex (component 3) by a red dashed-dotted line.



Figure 4. Concentration profile C(:,2) of the acyl complex normalized to maximum 1 by application of the simplified complementarity theorem in the form of algorithm 1. Left ordinate shows the nonscaled concentration resulting from algorithm 1; the right ordinate shows the absolute concentration of the acyl complex by using a kinetic model.³⁶

other factor. The mathematical background and the proofs of the complementarity and coupling theorems have been presented in ref12.

However, it is not evident how these theorems can practically be applied to spectroscopic data. The current work fills this gap and makes available short programs in MATLAB which can easily be applied and adapted to the needs of the users. The application of the software and the interpretation of its results have been explained step-by-step. The usefulness of the



Figure 5. Construction of the affine linear space for C(:,1). All linear combinations of the particular solution $c_p = U\Sigma W(:,1)$ (solid line) and the homogeneous solutions $c_h = U\Sigma y$ (broken line) span the affine subspace $C^{(1)}$ as given in (3). Then $C(:,1) = c_p + \gamma c_h$ for feasible values of γ .



Figure 6. Olefin component: Feasible nonscaled concentration profiles C(:,1) according to the coupling theorem and with $\gamma \in [a_1,b_1] = [1.19,1.98]$.

software is demonstrated for FT-IR spectroscopic data from the rhodium-catalyzed hydroformylation process.

Something which is not considered in this work is the socalled area of feasible solutions (AFS) and its combination with the complementarity theory. The simultaneous representation of all feasible nonnegative solutions in the form of a spectral AFS and a concentrational AFS is a very helpful and intuitive user interface for the application of the complementarity theory. For further details see refs 11, 26, and 37.







Figure 8. Complete factorization of the hydroformylation reaction system (top, concentration profiles; bottom, pure component spectra): blue solid line, olefin component; green broken line, the acyl complex; red dashed—dotted line, hydrido complex.

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Notes

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